

and XE-60.  $^1\text{H}$  and  $^{19}\text{F}$  NMR spectra were recorded on a Bruker CXP-200 spectrometer relative to  $\text{SiMe}_4$  and  $\text{CF}_3\text{COOH}$ , respectively.

**General procedure.** A solution of difluoride **1a–d** ( $4 \cdot 10^{-4}$  mol) in dry  $\text{CH}_2\text{Cl}_2$  (4–5 mL) was mixed with thoroughly dried KF ( $1 \cdot 10^{-4}$  mol) and 18-crown-6 ( $1 \cdot 10^{-4}$  mol). Then acetylene **2** ( $8 \cdot 10^{-4}$  mol) was added dropwise with stirring and cooling with ice water. The mixture was stirred for 0.5 h at  $-20^\circ\text{C}$  and analyzed by GLC. In all cases, only trace amounts of the initial acetylene **2** were observed in the reaction mixture, which indicates that the reaction was quantitatively. If necessary, the products were isolated by column chromatography (**5a,b**) on  $\text{SiO}_2$  or by preparative GLC.

Products **4**<sup>6</sup> and **6**<sup>7</sup> were synthesized by the known procedures. **5a**.  $^1\text{H}$  NMR,  $\delta$ : 7.46 (m).  $^{19}\text{F}$  NMR,  $\delta$ : -2.56 (m, 3 F); -20.5 (m, 2 F); -49.3 (m, 2 F). **5b**.  $^1\text{H}$  NMR,  $\delta$ : 7.42 (m, 2 H); 7.58 (m, 3 H).  $^{19}\text{F}$  NMR,  $\delta$ : -59.1 (m, 2 F); -76.1 (m, 1 F); -84.6 (m, 2 F).

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## Acylation of phenols with $\gamma$ -chlorobutyryl chloride and transformations of the reaction products

N. N. Yusubov

Baku State University,  
23 ul. P. Lumumby, 370602 Baku, Azerbaidzhan

Alkylphenols afford only *O*-acyl derivatives on treatment with  $\gamma$ -chlorobutyryl chloride in the presence of both  $\text{Et}_3\text{N}$  and  $\text{AlCl}_3$  at  $20$ – $60^\circ\text{C}$ . They cyclize under the action of  $\text{K}_2\text{CO}_3$  in DMSO into the respective cyclopropanes and undergo Fries rearrangement on heating with  $\text{AlCl}_3$  at  $120^\circ\text{C}$  into *C*-acyl derivatives.

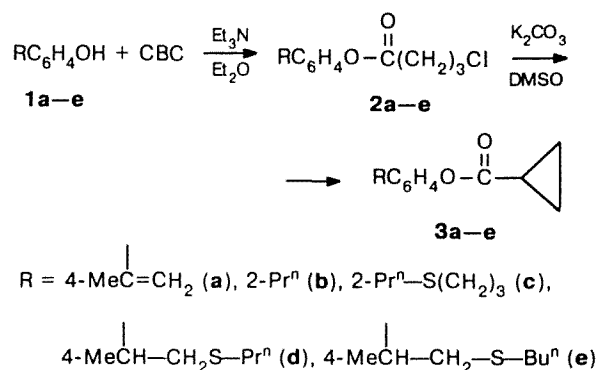
**Key words:** alkylphenols,  $\gamma$ -chlorobutyryl chloride, *O*-acylation, cyclopropanes, Fries rearrangement.

Phenol has previously been shown to undergo *O*-acylation on treatment with  $\gamma$ -chlorobutyryl chloride (CBC) in the presence of  $\text{Et}_3\text{N}$ . Under Friedel–Crafts conditions (in the presence of  $\text{AlCl}_3$ ), only 4-acyl derivatives are formed from alkoxybenzene and only 2-acyl derivatives are formed from 4-substituted alkoxybenzenes (4-OR, Br).<sup>1</sup> Some reactions of these compounds have been studied.<sup>2</sup>

In the present work, the *O*-acylation of phenols (**1a–e**)<sup>3–6</sup> on treatment with CBC in the presence of  $\text{Et}_3\text{N}$  and subsequent cyclization into cyclopropanes are carried out (Scheme 1).

Compounds **2d,e** were also synthesized by the thiylolation of *O*-acyl derivative **2a** in the presence of azobis-

Scheme 1

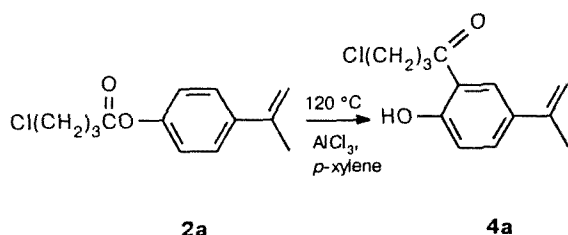


isobutyronitrile (AIBN). This reaction proceeds faster and in higher yield (90 % in 3.5 h at 80 °C) in comparison with the thiylation of original phenol **1a** (70 % in 25 h at 80 °C).<sup>4</sup>

The interaction of phenol **1d** with CBC under Friedel—Crafts conditions (in the presence of AlCl<sub>3</sub> at 20–60 °C) results only in the product of *O*-acylation **2d** in 20–30 % yield.

The products of *C*-acylation **4a–d** could be obtained as a result of Fries rearrangement in 32–57 % yields on heating ethers **2a–d** with AlCl<sub>3</sub> in *p*-xylene (3 h at 120 °C) according to Scheme 2 exemplified with product **4a**.

Scheme 2



The structure of new compounds was confirmed by <sup>1</sup>H NMR spectroscopy. Elemental analysis data are consistent with the proposed structures.

Polyfunctional phenols **4a–d** are of interest as new synthones for heterocycles.

### Experimental

<sup>1</sup>H NMR spectra were recorded in CCl<sub>4</sub> on a Tesla BS-487C (80 MHz) spectrometer. Starting phenols **1a–e** were prepared according to the known procedures.<sup>3–6</sup>

**O-(γ-Chlorobutyryl)-4-isopropenylphenol (2a).** CBC (14 g, 0.1 mol) was added dropwise to a mixture of 4-isopropenylphenol (13.4 g, 0.1 mol) and Et<sub>3</sub>N (10.1 g, 0.1 mol) in ether (60 mL) with stirring and cooling (–10 °C) over 1 h. The mixture was stirred for an additional 2 h, the precipitate was filtered off, and the solution was washed successively with alkali and water and dried with CaCl<sub>2</sub>. Then ether was evaporated, and the residue was distilled *in vacuo* to give product **2a** (16.4 g, 70 %) with b.p. 150 °C (1 Torr), *n*<sub>D</sub><sup>20</sup> 1.5368, *d*<sub>4</sub><sup>20</sup> 1.0997. Calculated (%): C, 65.41; H, 6.29; Cl, 14.88. C<sub>13</sub>H<sub>15</sub>ClO<sub>2</sub>. Found (%): C, 65.51; H, 6.34; Cl, 14.91. <sup>1</sup>H NMR, δ: 1.95 (m, 2 H, CCH<sub>2</sub>C); 2.00 (s, 3 H, Me); 2.52 (t, 2 H, CH<sub>2</sub>CO); 3.42 (t, 2 H, CH<sub>2</sub>Cl); 4.50 and 5.12 (m, 2 H, CH<sub>2</sub>=); 6.66 and 6.99 (m, 4 H, C<sub>6</sub>H<sub>4</sub>).

**O-(γ-Chlorobutyryl)-2-propylphenol (2b)** was synthesized similarly in 76 % yield, b.p. 228 °C (2 Torr), *n*<sub>D</sub><sup>20</sup> 1.5411, *d*<sub>4</sub><sup>20</sup> 1.1124. Found (%): C, 67.71; H, 7.21; Cl, 13.52. C<sub>13</sub>H<sub>17</sub>ClO<sub>2</sub>. Calculated (%): C, 67.54; H, 7.13; Cl, 13.32.

**O-(γ-Chlorobutyryl)-2-(3-propylthiopropyl)phenol (2c)** was synthesized similarly in 45.5 % yield, b.p. 205–208 °C (2 Torr), *n*<sub>D</sub><sup>20</sup> 1.5240, *d*<sub>4</sub><sup>20</sup> 1.11398. Found (%): C, 60.98; H, 7.04; Cl, 11.17; S, 9.88. C<sub>16</sub>H<sub>23</sub>ClO<sub>2</sub>S. Calculated (%): C, 61.05; H, 7.31; Cl, 11.29; S, 10.17. <sup>1</sup>H NMR, δ: 0.88 (t,

3 H, Me–CH<sub>2</sub>); 1.47 (m, 2 H, CH<sub>2</sub>–Me); 1.60 (m, 2 H, CH<sub>2</sub>–CH<sub>2</sub>–Ar); 1.75 (m, 2 H, CH<sub>2</sub>–CH<sub>2</sub>–Cl); 2.05 (t, 2 H, CH<sub>2</sub>CO); 2.35 (t, 4 H, CH<sub>2</sub>SCH<sub>2</sub>); 2.64 (m, 2 H, CH<sub>2</sub>Ar); 3.51 (t, 2 H, CH<sub>2</sub>Cl); 6.98 (m, 4 H, C<sub>6</sub>H<sub>4</sub>).

**O-(γ-Chlorobutyryl)-4-(1-methyl-2-propylthioethyl)phenol (2d)** was synthesized similarly in 69.5 % yield, b.p. 190–192 °C (3 Torr), *n*<sub>D</sub><sup>20</sup> 1.5315, *d*<sub>4</sub><sup>20</sup> 1.1322. Found (%): C, 61.27; H, 7.51; Cl, 11.12; S, 10.23. C<sub>16</sub>H<sub>23</sub>ClO<sub>2</sub>S. Calculated (%): C, 61.05; H, 7.31; Cl, 11.29; S, 10.17. <sup>1</sup>H NMR, δ: 0.84 (t, 3 H, Me); 1.16 (d, 3 H, Me); 1.34 (m, 2 H, CH<sub>2</sub>Me); 1.89 (m, 2 H, CH<sub>2</sub>CH<sub>2</sub>Cl); 2.30 (d, 2 H, SCH<sub>2</sub>CH); 2.54 (t, 2 H, CH<sub>2</sub>CO); 2.63 (t, 2 H, CH<sub>2</sub>S); 2.63 (m, 1 H, CH); 3.40 (t, 2 H, CH<sub>2</sub>Cl); 7.00 (m, 4 H, C<sub>6</sub>H<sub>4</sub>).

**O-(γ-Chlorobutyryl)-4-(2-butylthio-1-methylethyl)phenol (2e)** was synthesized similarly in 65.0 % yield, b.p. 195–198 °C (2 Torr), *n*<sub>D</sub><sup>20</sup> 1.5285, *d*<sub>4</sub><sup>20</sup> 1.0752. Found (%): C, 62.31; H, 7.72; Cl, 10.97; S, 9.64. C<sub>17</sub>H<sub>25</sub>ClO<sub>2</sub>S. Calculated (%): C, 62.10; H, 7.61; Cl, 10.80; S, 9.74. <sup>1</sup>H NMR, δ: 0.88 (t, 3 H, Me); 1.16 (d, 3 H, Me); 1.43 (m, 2 H, CH<sub>2</sub>Me); 1.58 (m, 2 H, CH<sub>2</sub>CH<sub>2</sub>S); 1.89 (m, 2 H, CH<sub>2</sub>CH<sub>2</sub>Cl); 2.30 (d, 2 H, CH<sub>2</sub>S); 2.54 (t, 2 H, CH<sub>2</sub>CO); 2.63 (t, 2 H, CH<sub>2</sub>S); 2.68 (m, 1 H, CH); 3.40 (t, 2 H, CH<sub>2</sub>Cl).

**O-Cyclopropylcarbonyl-4-isopropenylphenol (3a).** A mixture of ester **2a** (2.7 g, 11 mmol) and K<sub>2</sub>CO<sub>3</sub> (1.57 g, 11 mmol) in DMSO (40 mL) was stirred for 2 h at 60–70 °C. The mixture was cooled, water (100 mL) was added, and the product was extracted with ether. The extract was dried with CaCl<sub>2</sub>, and after evaporation of ether, the residue was distilled *in vacuo* to give product **3a** (2.0 g, 87 %), b.p. 120 °C (0.2 Torr), *n*<sub>D</sub><sup>20</sup> 1.5470, *d*<sub>4</sub><sup>20</sup> 1.5470. Found (%): C, 77.18; H, 6.95. C<sub>13</sub>H<sub>14</sub>O<sub>2</sub>. Calculated (%): C, 77.23; H, 6.93. <sup>1</sup>H NMR, δ: 0.44 and 0.72 (m, 4 H, 2 CH<sub>2</sub> of the cycle); 1.58 (m, 1 H, CH); 2.00 (s, 3 H, Me); 4.49 and 5.14 (m, 2 H, =CH<sub>2</sub>); 6.66 and 6.99 (m, 4 H, C<sub>6</sub>H<sub>4</sub>).

**O-Cyclopropylcarbonyl-2-propylphenol (3b)** was synthesized similarly in 82.6 % yield, b.p. 147–150 °C (4 Torr), *n*<sub>D</sub><sup>20</sup> 1.5311, *d*<sub>4</sub><sup>20</sup> 1.09714. Found (%): C, 76.39; H, 7.69. C<sub>13</sub>H<sub>16</sub>O<sub>2</sub>. Calculated (%): C, 76.47; H, 7.84.

**O-Cyclopropylcarbonyl-2-(3-propylthiopropyl)phenol (3c)** was synthesized similarly in 64.95 % yield, b.p. 177–180 °C (4 Torr), *n*<sub>D</sub><sup>20</sup> 1.5250, *d*<sub>4</sub><sup>20</sup> 1.09205. Found (%): C, 68.87; H, 7.64; S, 11.03. C<sub>16</sub>H<sub>22</sub>O<sub>2</sub>S. Calculated (%): C, 69.06; H, 7.91; S, 11.51. <sup>1</sup>H NMR, δ: 0.44 and 0.72 (m, 4 H, 2 CH<sub>2</sub> of the cycle); 0.87 (t, 3 H, Me); 1.47 (m, 2 H, CH<sub>2</sub>Me); 1.58 (m, 1 H, CH); 1.74 (m, 4 H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>); 2.39 (t, 4 H, CH<sub>2</sub>SCH<sub>2</sub>); 2.61 (t, 2 H, CH<sub>2</sub>Ar); 6.91 (m, 4 H, C<sub>6</sub>H<sub>4</sub>).

**O-Cyclopropylcarbonyl-4-(1-methyl-2-propylthioethyl)phenol (3d)** was synthesized similarly in 84.7 % yield, b.p. 160–165 °C (1 Torr), *n*<sub>D</sub><sup>20</sup> 1.5420, *d*<sub>4</sub><sup>20</sup> 1.1322. Found (%): C, 69.21; H, 8.02; S, 11.70. C<sub>16</sub>H<sub>22</sub>O<sub>2</sub>S. Calculated (%): C, 69.06; H, 7.91; S, 11.51. <sup>1</sup>H NMR, δ: 0.44 and 0.72 (m, 4 H, 2 CH<sub>2</sub> of the cycle); 0.85 (t, 3 H, Me); 1.16 (d, 3 H, MeCH); 1.40 (m, 2 H, CH<sub>2</sub>Me); 1.58 (m, 1 H, CH); 2.3 (d, 2 H, CH<sub>2</sub>S); 2.63 (t, 2 H, SCH<sub>2</sub>); 2.68 (m, 1 H, CH); 6.70 and 7.00 (m, 4 H, C<sub>6</sub>H<sub>4</sub>).

**O-Cyclopropylcarbonyl-4-(2-butylthio-1-methylethyl)phenol (3e)** was synthesized similarly in 75.5 % yield, b.p. 180–183 °C (4 Torr), *n*<sub>D</sub><sup>20</sup> 1.5428, *d*<sub>4</sub><sup>20</sup> 1.0817. Found (%): C, 69.88; H, 8.19; S, 10.97. C<sub>17</sub>H<sub>24</sub>O<sub>2</sub>S. Calculated (%): C, 69.82; H, 8.22; S, 10.96.

**2-(γ-Chlorobutyryl)-4-isopropenylphenol (4a).** A mixture of compound **2a** (0.1 mol) and AlCl<sub>3</sub> (0.1 mol) in *p*-xylene (200 mL) was stirred for 3 h at 120 °C in an argon atmosphere. After cooling, the mixture was poured into a solution of conc. HCl (20 mL) in water (800 mL). The organic layer was separated and washed with water to pH 7, dried with

MgSO<sub>4</sub>, and concentrated, and the residue was distilled *in vacuo* to afford **4a** (3.2 g, 56.8 %) with b.p. 150–151 °C (2 Torr),  $n_D^{20}$  1.5271,  $d_4^{20}$  1.2583. Found (%): C, 65.5; H, 6.30; Cl, 14.91. C<sub>13</sub>H<sub>15</sub>ClO<sub>2</sub>. Calculated (%): C, 65.14; H, 6.29; Cl, 14.88. <sup>1</sup>H NMR,  $\delta$ : 1.58 (m, 2 H, CCH<sub>2</sub>C); 1.70 (s, 3 H, Me); 2.04 (t, 2 H, CH<sub>2</sub>CO); 3.51 (t, 2 H, CH<sub>2</sub>Cl); 4.60 and 5.10 (m, 2 H, CH<sub>2</sub>=); 6.51 (br.s, 1 H, OH); 6.94 (m, 3 H, C<sub>6</sub>H<sub>3</sub>).

**2-( $\gamma$ -Chlorobutyryl)-6-propylphenol (4b)** was synthesized similarly in 50 % yield, b.p. 176–180 °C (1.5 Torr),  $n_D^{20}$  1.5047,  $d_4^{20}$  1.0878. Found (%): C, 65.61; H, 6.32; Cl, 14.96. C<sub>13</sub>H<sub>17</sub>ClO<sub>2</sub>. Calculated (%): C, 65.41; H, 6.29; Cl, 14.88.

**2-( $\gamma$ -Chlorobutyryl)-6-(3-propylthiopropyl)phenol (4c)** was synthesized similarly in 31.9 % yield, b.p. 208–210 °C (1.5 Torr),  $n_D^{20}$  1.5290,  $d_4^{20}$  1.1100. Found (%): C, 60.97; H, 7.04; Cl, 11.16; S, 9.89. C<sub>16</sub>H<sub>22</sub>ClO<sub>2</sub>S. Calculated (%): C, 61.05; H, 7.31; Cl, 11.29; S, 10.17. <sup>1</sup>H NMR,  $\delta$ : 0.88 (t, 3 H, Me); 1.43 (m, 2 H, CH<sub>2</sub>Me); 1.58 (m, 2 H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>); 1.76 (m, 2 H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>); 2.04 (t, 2 H, CH<sub>2</sub>CO); 2.34 (t, 4 H, CH<sub>2</sub>SCH<sub>2</sub>); 2.66 (t, 2 H, CH<sub>2</sub>Ar); 3.51 (t, 2 H, CH<sub>2</sub>Cl); 6.51 (br.s, 1 H, OH); 6.94 (m, 3 H, C<sub>6</sub>H<sub>3</sub>).

**2-( $\gamma$ -Chlorobutyryl)-4-(1-methyl-2-propylthioethyl)phenol (4d)** was synthesized similarly in 46 % yield, b.p. 180–183 °C (4 Torr),  $n_D^{20}$  1.5401,  $d_4^{20}$  1.1354. Found (%): C, 56.27; H, 8.17; Cl, 13.01; S, 11.54. C<sub>16</sub>H<sub>23</sub>ClO<sub>2</sub>S. Calculated (%): C, 56.01; H, 8.26; Cl, 12.75; S, 11.49. <sup>1</sup>H NMR,  $\delta$ : 0.84 (t,

3 H, Me); 1.16 (q, 3 H, Me); 1.43 (m, 2 H, CH<sub>2</sub>Me); 1.90 (m, 2 H, CCH<sub>2</sub>C); 2.04 (t, 2 H, CH<sub>2</sub>CO); 2.34 (t, 4 H, CH<sub>2</sub>SCH<sub>2</sub>); 2.68 (m, 1 H, CH); 3.51 (t, 2 H, CH<sub>2</sub>Cl); 6.51 (br.s, 1 H, OH); 6.94 (m, 3 H, C<sub>6</sub>H<sub>3</sub>).

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## Reaction of phenoxazine and phenothiazine with 1,1-dicyano-2-(trifluoromethyl)ethylenes

A. V. Fokin, A. Yu. Sizov, V. I. Dyachenko,\* V. D. Sviridov, and N. D. Chkanikov

A. N. Nesmeyanov Institute of Organoelement Compounds, Russian Academy of Sciences, 28 ul. Vavilova, 117813 Moscow, Russian Federation.

Fax: +7 (095) 135 5085

1,1-Dicyano-2,2-bis(trifluoromethyl)ethylene alkylates phenoxazine and phenothiazine at 20 °C at the *para*-position relative to the N atom.

**Key words:** phenoxazine, phenothiazine, 1,1-dicyano-2,2-bis(trifluoromethyl)ethylene, methyl 3,3-dicyano-2-(trifluoromethyl)acrylate, C-alkylation.

Reactions of phenoxazine and phenothiazine with tetracyanoethylene in DMF at 100 °C give the products of tricyanovinylolation at the *para*-position relative to the N atom as a result of abstraction of HCN from the initially formed C-alkylation products.<sup>1</sup>

It is known that 1,1-dicyano-2,2-bis(trifluoromethyl)ethylene (**1**) and esters of 3,3-dicyano-2-(trifluoromethyl)acrylic acid can C-alkylate electron-donor

aromatic and heteroaromatic compounds under mild conditions.<sup>2</sup>

In the present work, the reactions of phenoxazine and phenothiazine with dicyanoethylene **1** and methyl 3,3-dicyano-2-(trifluoromethyl)acrylate (**2**) were studied.

Phenoxazine and phenothiazine appeared to undergo C-alkylation by dicyanoethylene **1** already at 20 °C. In